Teri L. Copeland, M.S., DABT Consulting Toxicologist

5737 Kanan Rd. #182, Agoura Hills, CA 91301 * (818) 991-8240 phone (818) 991-8140 fax

Technical Memorandum

То:	Brian A. Rakvica, P.E. Nevada Division of Environmental Protection, Bureau of Corrective Actions
From:	Teri Copeland, M.S., DABT and Joanne Otani Fehling, R.N., M.S.N., P.H.N
Date:	February 6, 2006
Re:	Selection of Pyrene as a Noncarcinogenic Toxicological Surrogate for PAHs

This technical memorandum provides a toxicity-based assessment to support the selection of a surrogate chemical for purposes of the assessment of potential noncancer toxicity associated with oral exposure to polycyclic aromatic hydrocarbons (PAHs) that have not been assigned a noncancer oral toxicity criterion by USEPA. These PAHs include:

PAHs Classified by USEPA as Carcinogens

Benzo(a)anthracene Benzo(a)pyrene Benzo(b)fluoranthene Benzo(k)fluoranthene Chrysene Dibenz(a,h)anthracene Indeno(1,2,3-c,d)pyrene

PAHs Classified by USEPA as Noncarcinogens

Acenaphthylene Benzo(g,h,i)perylene Phenanthrene

I. Methodology and Analysis

A toxicological surrogate was selected from a list of candidate chemicals (i.e., PAHs for which noncancer oral toxicity criteria have been assigned by USEPA) based on consideration of the following:

- Availability of relevant toxicity data;
- Target organs;
- Dose-response information; and
- Structure-activity considerations.

Additionally, physical-chemical properties were considered, although they relate more to environmental fate and transport than to toxicity. Many of the physical-chemical properties for the carcinogenic PAHs are similar to the selected toxicological surrogate, pyrene (within an order magnitude), with the exception of the vapor pressures and Henry's Law Constants.

A summary of information relied upon in the selection of the toxicological surrogate is presented in Tables 1 and 2.

Availability of Relevant Toxicity Data

Relevant oral toxicity data were reported by USEPA for six noncarcinogenic PAHs (acenaphthene, anthracene, fluoranthene, fluorene, naphthalene, and pyrene). Information was taken from the USEPA Integrated Risk Information System (IRIS, USEPA, 2006), the Toxicological Profile for Polycyclic Aromatic Hydrocarbons (ATSDR, 1990; 1995), and the Toxicity Summary for Pyrene (ORNL, 1993) and supporting documentation contained in those data sources. The primary toxicity studies for all six PAHs were oral gavage, subchronic (90-day) rodent studies. Relevant summaries are provided in Table 1.

Target Organs

For the six candidate PAHs,, the most sensitive target organs for noncancer toxicity were identified as liver, kidney, and blood (hematological effects) (see Table 1).

For the seven carcinogenic PAHs, target organs for noncancer toxicity include liver, kidney, and blood. Relevant data provided by USEPA (2006), ATSDR, 1995, and ORNL (1993) include:

- A single intragastric administration of benzo(a)pyrene (200 mg/kg-day), benzo(a)anthracene (200 mg/kg-day), or dibenz(a,h)anthracene (180 mg/kg-day) resulted in the induction of foci of altered hepatocytes¹ (Tsuda and Farber, 1980, cited in ATSDR, 1995);
- Benzo(a)pyrene, benzo(a)anthracene, and chrysene (as well as anthracene and phenanthrene) induced the liver enzyme aldehyde dehydrogenase following short-term intragastric administration of 100 mg/kg-day (Torronen et al., 1981, cited in ATSDR, 1995);
- Oral exposure to benzo(a)pyrene and benzo(a)anthracene increased relative liver weights in rats by 27% and 19%, respectively² (Torronen et al., 1981, cited in ATSDR, 1995);
- Benzo(a)pyrene, benzo(a)anthracene, and chrysene were moderate inducers of the enzyme carboxylase in the liver of rats that were intragastrically administered doses of 50 150 mg/kg-day (Nousiainen et al., 1984, cited in ATSDR, 1995);
- Mice that were orally administered benzo(a)pyrene at 120 mg/kg-day in the diet for 180 days exhibited a 13% increase in relative liver weights (Robinson et al., 1975, cited in ATSDR, 1995);

¹ Liver cells.

² Doses were not given in the ATSDR summary of this study.

- Benzo(a)pyrene induced the enzyme carboxylase in the kidney of rats that were intragastrically administered doses of 50 150 mg/kg-day (Nousiainen et al., 1984, cited in ATSDR, 1995);
- In a 13-week oral study, mice were administered pyrene in corn oil at doses of 0, 75, 125, or 250 mg/kg-day. Dose-related nephropathy³ was observed in females; nephropathy was seen in males at the high dose (250 mg/kg-day) only. Absolute and relative kidney weights were reduced in the two highest dose groups. Relative liver weights were increased in males exposed to 250 mg/kg-day and in females exposed to 125 or 250 mg/kg-day. Treated males exhibited minor hematological changes that consisted of decreased red blood cell count, packed cell volume, and hemoglobin levels (Toxicity Research Laboratories, 1989, cited in ORNL, 1993);
- Adverse hematopoietic⁴ effects were reported in mice following oral exposure to 120 mg/kg-day benzo(a)pyrene for 180 days (Robinson et al., 1975, cited in ATSDR, 1995).

Dose-Response Information

The no-observed-adverse-effect-level (NOAEL) for the six candidate PAHs ranged from 71 mg/kg-day (naphthalene) to 1,000 mg/kg-day (anthracene) (USEPA, 2006). The lowest-observed-adverse-effect-level (LOAEL) for the six candidate PAHs ranged from 125 mg/kg-day (pyrene) to 350 mg/kg-day (acenaphthalene). A LOAEL was not identified for anthracene, as no adverse effects were observed in the studies reviewed by USEPA (2006), however it is reasonable to conclude that a LOAEL for anthracene would be greater than the NOAEL of 1,000 mg/kg-day. For the carcinogenic PAHs, reported toxicological thresholds for noncancer effects are generally in the range of those for the noncancer PAHs (ATSDR, 1995).

The USEPA oral reference dose $(RfD_o)^5$ ranges from 0.3 mg/kg-day (anthracene) to 0.02 mg/kg-day (naphthalene). The RfD_o for naphthalene is based on a decrease in body weight rather than a specific organ effect.

Structure-Activity Considerations

PAHs are a class of structurally similar compounds characterized by the presence of fused aromatic rings (see Tables 1 and 2 for structures). As shown in Table 1, the six candidate noncarcinogenic PAHs consist of two to four rings. The seven carcinogenic PAHs consist of four to six rings. Based on the noncarcinogenic toxicity endpoints and ranges of noncarcinogenic threshold effect levels (for noncarcinogenic as well as carcinogenic PAHs) reported by USEPA (2006), ATSDR (1995), and ORNL (1993), it can be inferred that there is some degree of structure-activity relationship for noncancer toxicity⁶, although the data are not adequate for quantitative assessment of structure-activity relationships.

³ Any disease of the kidney.

⁴ Pertaining to or related to the formation of blood cells.

⁵ In general, the RfD is an estimate (with uncertainty spanning perhaps an order of magnitude) of a daily exposure to the human population (including sensitive subgroups) that is likely to be without an appreciable risk of deleterious effects during a lifetime (USEPA, 2006).

⁶ The structure-activity relationship for the carcinogenic PAHs is well documented (USEPA, 1993, 2006; ATSDR, 1995). ATSDR (1995) contends that PAH toxicity generally increases with the number of rings, however some exceptions are found in the available toxicity data.

II. Selection of Noncancer Toxicological Surrogate for Oral Exposure Routes

Pyrene was selected as the toxicological surrogate for purposes of characterizing potential noncancer risks (i.e., hazard quotients) for PAHs not assigned an oral reference dose by USEPA. This selection is supported by the following rationale:

Availability of Relevant Toxicity Data

Relevant oral toxicity data were reported by USEPA for acenaphthene, anthracene, fluoranthene, fluorene, naphthalene, and pyrene.

Target Organs

For the seven carcinogenic PAHs, target organs for noncancer toxicity include liver, kidney, and blood. For the six candidate PAHs, the most sensitive target organs for noncancer toxicity were identified as follows:

- None Anthracene
- Kidney, liver, and hematological Fluoranthene and fluorene
- Kidney and hematological Pyrene
- Liver Acenaphthalene
- Body weight (decreased) and possible hematological Naphthalene

The RfD_o for naphthalene is based on a decrease in terminal body weight. It should be noted that hematological effects have also been reported, although USEPA questions the toxicological significance of those reported effects. According to the USEPA, the decrease in terminal body weight was considered the most significant and sensitive endpoint (USEPA, 1996).

Based on target organ information, flouranthene, fluorene, naphthalene, and pyrene are considered to be the best surrogate candidates.

Dose-Response Information

For the carcinogenic PAHs, reported toxicological thresholds for noncancer effects are generally in the range of those for the noncancer PAHs (ATSDR, 1995). The lowest LOAELs for the six candidate PAHs were reported at 125 mg/kg-day for pyrene and 142 mg/kg-day for naphthalene (USEPA, 2006). The lowest (most stringent) RfD_os assigned by USEPA are 0.02 mg/kg-day for naphthalene and 0.03 mg/kg-day for pyrene.

Based on dose-response information, naphthalene and pyrene are considered to be the best surrogate candidates.

Structure-Activity Considerations

The two surrogate candidates that exhibit the lowest thresholds for noncancer effects are naphthalene and pyrene. The pyrene structure is more closely associated with the structures of the carcinogenic PAHs, which contain four to six rings. ATSDR (1995) contends that PAH toxicity generally increases with the number of rings. However, for noncancer toxicity, there has been no formal analysis to support a broad application of that position.

Based on structure-activity considerations, pyrene is considered to be the best surrogate candidate.

Summary

Fluoranthene, flourene, naphthalene, and pyrene represent acceptable surrogate choices based on toxicity considerations (e.g., dose-response relationships, effect levels, and USEPA RfD_0). Of these four candidates, naphthalene and pyrene exhibit slightly higher noncancer toxicity compared with fluoranthene and fluorene. Pyrene was ultimately selected over naphthalene due to (1) noncancer toxicity endpoints are more consistent with those for carcinogenic PAHs and (2) the greater number of rings in the pyrene chemical structure.

III. References Cited

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Parameters	Acenaphthene	Anthracene	Fluoranthene	Fluorene Naphthalen		Pyrene
Structure						
CAS No.	83-29-9	120-12-7	206-44-0	86-73-7	91-20-3	129-00-00
Formula	C ₁₂ H ₁₀	$C_{14}H_{10}$	$C_{16}H_{10}$	C ₁₃ H ₁₀	C10H8	$C_{16}H_{10}$
MW (g/mole)	154.21	178.2	202.26	166.2	128.19	202.3
MP/BP (°C)	95 /96.2	218/340, 342	11/375	116-117 / 295	80.5/218	156 /393, 404
H ₂ O Solubility (mg/L)	1.93	0.076	0.20-0.26	1.68-1.98	31.7	0.077
Log Kow/Koc	3.98/3.66	4.45/4.15	4.90/4.58	4.18/3.86	3.29/ 2.97	4.88/4.58
VP (mmHg)	<u>4.47E-03</u>	<u>1.7E-05 @25 °C</u>	5.0E-06 @25 °C	3.2E-04 @20 °C	8.7E-02 @25 °C	2.5E-06 @25 °C
HLC (atm-m ³ /mol)	7.91E-05	1.77E-05	6.5E-06	1.0E-04	4.6E-04	1.14E-05
Oral RfD (mg/kg-day)	6E-02	3E-01	4E-02	4E-02	2E-02	3E-02
Study Length	90 days	90 days	13 weeks	13 weeks	13 weeks	13 weeks
Species/Route	Mouse/gavage	Mouse/gavage	Mouse/gavage	Mouse/gavage	Rat/gavage	Mouse/gavage
Organ	Liver	None	Kidney/Liver/Hematological	Kidney/Liver/Hematological	Body weight/Hematological	Kidney/Hematological
Toxicity Endpoint(s)) fi liver wt, dose-	None observed at doses of	Dose-dependent îî liver	↓ RBC packed cell volume	\Downarrow mean terminal body wt in	Renal tubular pathology
	dependent cellular	0, 250, 500, and 1,000	enzymes, nephropathy, and	and Hemoglobin (both	males (30% and 12% at high	characterized by multiple foci
	hypertrophy, sign. î	mg/kg-day (USEPA,	salivation; 们 liver wt and	males/females at 250 mg/kg-	and mid-dose levels,	tubular regeneration often
	cholesterol @ mid-high	1989b as cited in IRIS)	hematological effects at 250	day); also observed sign. î	respectively) considered most	accompanied by interstitial
	doses (350, 750 mg/kg-		and 500 mg/kg-day (USEPA,	liver wt at 250/500; ft	sensitive endpoint. ↓ liver,	lymphocytic infiltrates +/-
	day) (USEPA, 1989a as	[1988 as cited in IRIS)	kidney/spleen wts in high dose	brain, spleen wts in high dose	interstitial fibrosis (dose-
	cited in IRIS)			males/females (USEPA, 1989c	females (133 mg/kg-day)	dependent in females); U
				as cited in IRIS)	(Shopp et al., 1984 as cited in	kidney wt. If liver wt in males
					USEPA 1998). Limited	(250 mg/kg/day) & females
		}			hematological effects at high	(≥125 mg/kg/day). Slight ↓
		}			dose (400 mg/kg-day). No	hematological parameters in
					dose-dependent changes in	males. (USEPA, 1989d as
				}	kidneys. (BCL, 1980 as cited	cited in IRIS; ORNL, 1993;
				·····	in IRIS)	ATSDR, 1995).
NUAEL	175 mg/kg-day	1,000 mg/kg-day	125 mg/kg-day	125 mg/kg-day	71 mg/kg-day	/5 mg/kg-day
LUAEL	350 mg/kg-day	None	250 mg/kg-day	250 mg/kg-day	142 mg/kg-day	125 mg/kg-day
	3000/1	3000/1	3000/1	3000/1	3000/1	3000/1
KID Confidence	Study – Low	Study – Low	Study – Medium	Study – Medium	Study – High	Study – Medium
(Study/Database/RfD)	Database – Low	Database – Low	Database – Low	Database – Low	Database – Low	Database – Low
	RTD - Low	RID - LOW	RfD - Low	KID - LOW	KID – Low	KID – LOW

Table 1: Summary of Chemical-Physical Properties and Oral Noncancer Toxicity Information for Noncarcinogenic Polycyclic Aromatic Hydrocarbons*

1

*No toxicity data are available for accnaphthylene, benzo(ghi)penylene, and phenanthrene. MW – Molecular weight; MP/BP – Melting point/Boiling point; Kow – Octanol water partition coefficient; Koc – Octanol carbon partition coefficient; VP – Vapor pressure; HLC – Henry's Law Constant; NOAEL/LOAEL – No observable adverse effect level/ Lowest observable adverse effect level; RfD – reference dose: UF/MF – uncertainty factor/ modifying factor; sign – statistically significant; wt – weight; (1 - Increase value; U) - Decrease value. References for Physical/Chemical Properties: ATSDR, 1990 (naphthalene only); ATSDR, 2005 (all other PAHs).

Parameters	Benzo(a)anthracene	Benzo(a)pyrene	Benzo(b)fluoranthene	Benzo(k)fluoranthene	Chrysene	Dibenz(a,h)anthracene	Indeno(1,2,3-cd)
							pryene
Structure							
CAS No.	56-55-3	50-32-8	205-99-2	207-08-9	218-01-9	53-70-3	193-39-5
Formula	$C_{18}H_{12}$	$C_{20}H_{12}$	C ₂₀ H ₁₂	$C_{20}H_{12}$	C ₁₈ H ₁₂	C ₂₂ H ₁₄	$C_{22}H_{12}$
MW (g/mole)	228.29	252.3	252.3	252.3	228.3	278.35	276.3
MP/BP (°C)	158-159; 162	179-179.3	168.3	215.7	255-256	262	163.6
H ₂ 0 Solubility (mg/L)	1.0E-02	2.3E-03	1.2E-03	7.6E-04	2.8E-03	5E-04	6.2E-02
Log Kow/Koc	5.61/5.30	6.06/6.74	6.04/5.74	6.06/5.74	5.16/5.30	6.84/6.52	6.58/6.20
VP (mmHg)	2.2E-08 @ 20 °C	5.6E-09	5.0E-07 @ 20-25 °C	9.59E-11	6.3E-07	1E-10	~10E-11 to 10E-06
HLC (atm-m ³ /mol)	1.0E-06	4.9E-07	1.22E-05	3.87E-05	1.05E-06	7.3E-08	6.95E-08

Table 2: Summary of Chemical-Physical Properties for Carcinogenic Polycyclic Aromatic Hydrocarbons

 ↑ - Bay region. Active component of molecule for carcinogenicity (ATSDR, 1995).

 ↑ - MW - Molecular weight; MP/BP - Melting point/Boiling point; Kow - Octanol water partition coefficient; Koc - Octanol carbon partition coefficient; VP - Vapor pressure; HLC - Henry's Law Constant References: ATSDR, 2005.